## AN ALTERNATIVE COMPUTATIONAL METHOD TO THAT OF JUDD

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Judd and Norris [1] have dealt with the determination of the reaction order n from the equation

$$(1 - \alpha)^{1 - n} = 1 + (n - 1)kt \tag{1}$$

They concluded that the only correct method is the application of the least squares principle

$$\sum_{i} (\delta \alpha_{i})^{2} = \min$$
 (2)

where the symbol  $\delta \alpha_i$  stands for the difference between the observed and the calculated values of  $\alpha_i$ .

Since Eq. (1) is non-linear by two parameters they solved the problem by a general non-linear minimization method.

One of the non-linear parameters, however, can be eliminated if we use the following procedure:

Suppose that n is fixed. Eq. (1) is then linear for k and can be written as

$$Y = 1 + (n - 1) kt$$
(3)

where

$$Y = (1 - \alpha)^{1 - n}$$
 (4)

If we differentiate Eq. (4), an arbitrary but not too great change of  $\alpha$  can be expressed as follows:

$$\delta \alpha \cong \frac{1}{1-n} (1-\alpha)^n \, \delta Y \tag{5}$$

Using relation (5), condition (2) can be well approximated:

$$\sum_{i} (\delta \alpha_{i})^{2} \cong \sum_{i} c_{i}^{2} (\delta Y_{i})^{2} = \min$$
(6)

where

$$c_{i} = \frac{1}{1 - n} (1 - \alpha)^{n}$$
(7)

J. Thermal Anal. 5, 1973

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For given *n* relations (3), (6) and (7) present together an extremely simple linear least squares problem. If the value of *n* is changed systematically the best value of *n* can easily be found. The simplest way is to calculate sum (6) for 10-15 different values of *n* and choose the least one.

## Reference

1. M. D. Judd and A. C. Norris, J. Thermal Anal., 5 (1973) 179.